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A Microscale Approach to Chemical Kinetics in the General Chemistry Laboratory: The Potassium Iodide Hydrogen Peroxide Iodine-Clock Reaction Prem D. Sattangi Department of Chemistry, The Pennsylvania State University, Fayette, The Eberly Campus, Uniontown, Pennsylvania 15401, United States

A Microscale Approach to Chemical Kinetics in the General

Microscale General Chemistry Laboratory with Select Macroscopic Experiments; Wiley: New York, 1993. Experiment 2 Dissolved Oxygen in Water Reference Chapters: 6, 7, 8 Objectives After performing this experiment, the student shall be able to: • Determine the level of dissolved oxygen in a sample of water using Winkler's method.

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Students who are going on in chemistry probably should have the exposure with burettes. Should a teacher feel "bad" or "guilty" if he or she does not get to the burettes? That depends...are we teaching the students to be chemists or to think like chemists? I would suggest the latter. Both large scale and microscale have advantages and drawbacks.

Titration and Microscale Chemistry | Chemical Education

Department of Chemistry GENERAL CHEMISTRY 101 LABORATORY MANUAL An Inquiry Approach through an Environmental Awareness The following laboratories have been compiled and adapted by Alan Khuu, M.S. & Armando Rivera, Ph.D.

GENERAL CHEMISTRY 101 LABORATORY MANUAL

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